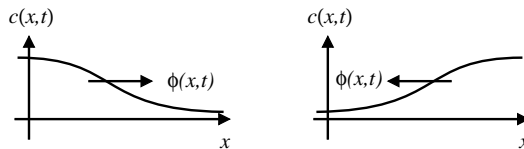
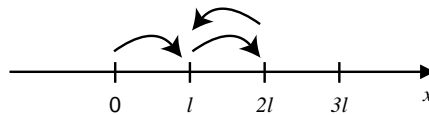


Review of Lecture 2

- Diffusion = transport of solute due to gradient of solute concentration
- Fick's first law:  $\phi(x,t) = -D \frac{\partial c(x,t)}{\partial x}$ 
  - inspired by experimental work of Graham
  - developed by analogy to laws for heat and charge transport



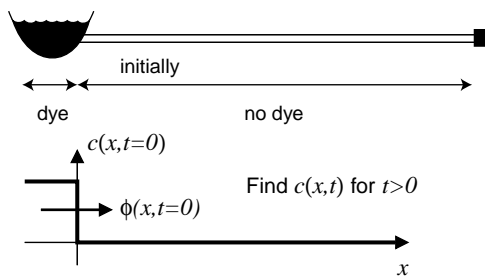
- Microscopic basis: Einstein's random walk model
  - predicts Brownian motion at a microscopic scale
  - predicts Fick's first law at a macroscopic scale



- Demonstration



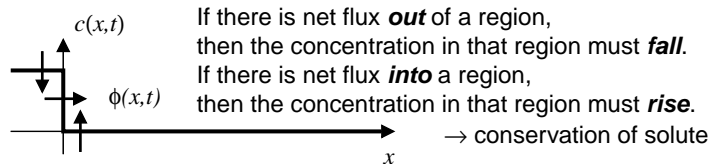
Apply Fick's law to dye demonstration



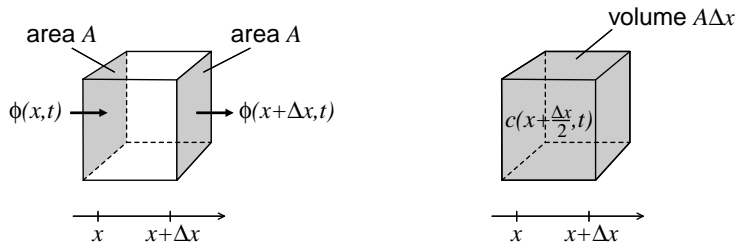
Fick's law:

$$\phi(x,t) = -D \frac{\partial c(x,t)}{\partial x}$$

- provides information about time "t" only
- need new information to get from time "t" to time "t+Δt"



### Continuity Equation



Amount of solute entering through edges during  $(t, t+\Delta t)$

Change in amount of solute in volume from  $t$  to  $t+\Delta t$

$$\text{--- left ---} \quad \text{--- right ---} \quad \text{--- time 2 ---} \quad \text{--- time 1 ---}$$

$$\phi(x, t + \frac{\Delta t}{2}) A \Delta t - \phi(x + \Delta x, t + \frac{\Delta t}{2}) A \Delta t = c(x + \frac{\Delta x}{2}, t + \Delta t) A \Delta x - c(x + \frac{\Delta x}{2}, t) A \Delta x$$

↑  
equal if solute is neither created nor destroyed

$$\frac{\phi(x, t + \frac{\Delta t}{2}) - \phi(x + \Delta x, t + \frac{\Delta t}{2})}{\Delta x} = \frac{c(x + \frac{\Delta x}{2}, t + \Delta t) - c(x + \frac{\Delta x}{2}, t)}{\Delta t}$$

Take limit as  $\Delta x \rightarrow 0$  and  $\Delta t \rightarrow 0$

$$-\frac{\partial \phi(x, t)}{\partial x} = \frac{\partial c(x, t)}{\partial t}$$

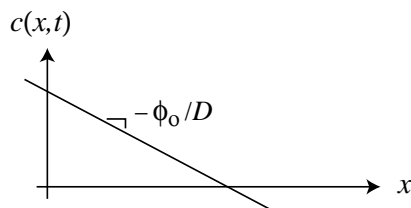
Fick's first law:  $\phi(x, t) = -D \frac{\partial c(x, t)}{\partial x}$

Continuity equation:  $-\frac{\partial \phi(x, t)}{\partial x} = \frac{\partial c(x, t)}{\partial t}$

Diffusion equation:  $\frac{\partial c(x, t)}{\partial t} = D \frac{\partial^2 c(x, t)}{\partial x^2}$

Time invariant solutions to diffusion equation:  
what happens when diffusion proceeds for a long time?

$$\begin{aligned} \frac{\partial \phi(x, t)}{\partial t} \rightarrow 0 & \quad \frac{\partial c(x, t)}{\partial t} \rightarrow 0 \Rightarrow \frac{\partial \phi(x, t)}{\partial x} \rightarrow 0 \\ & \searrow \quad \swarrow \\ & \phi(x, t) \rightarrow \text{constant} \\ & \Downarrow \\ -D \frac{\partial c(x, t)}{\partial x} & = \text{constant} = \phi_0 \\ & \Downarrow \\ c(x, t) & = \text{linear function of } x = c(x, t_0) \end{aligned}$$



If no external sources and sinks → closed system  
 then sustained fluxes are not possible (require energy)

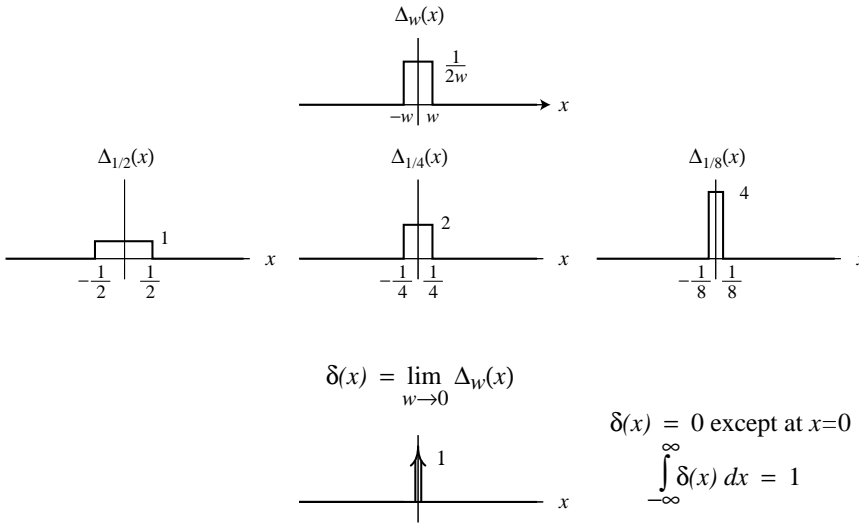
⇒ Equilibrium: persistent state of zero flux

$$\begin{aligned} \Downarrow \qquad \qquad \Downarrow \\ \frac{\partial c(x,t)}{\partial t} = 0 \qquad \phi(x,t) = 0 \\ \frac{\partial \phi(x,t)}{\partial t} = 0 \end{aligned}$$

$$\phi(x,t) = -D \frac{\partial c(x,t)}{\partial x} = 0$$

$\nearrow$   $D = 0$  (not diffusible)       $\nwarrow$   $c(x,t) = c_0$  (constant concentration)

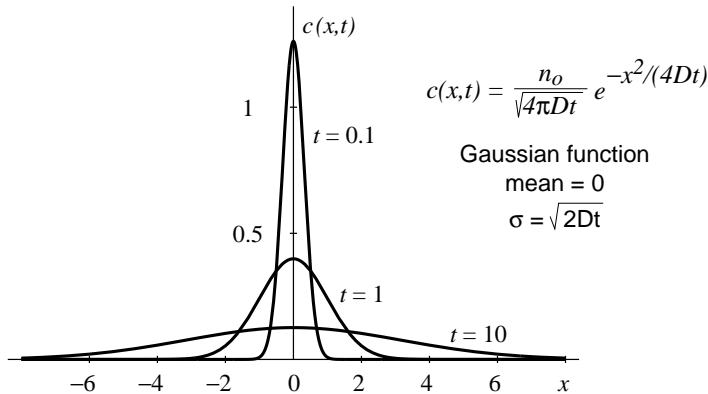
Dirac delta function  $\delta(x)$



Impulse Response

Given  $c(x,t) = n_0 \delta(x)$  at  $t=0$

Solve  $\frac{\partial c(x,t)}{\partial t} = D \frac{\partial^2 c(x,t)}{\partial x^2}$  for  $t > 0$



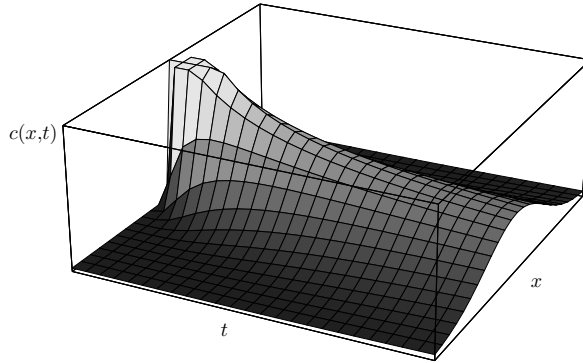
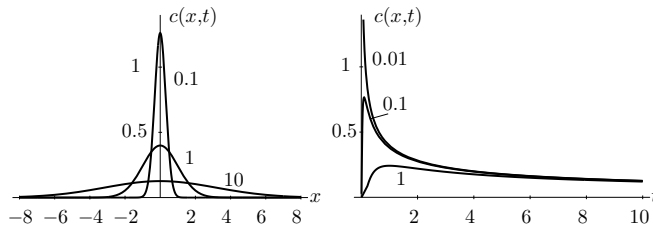
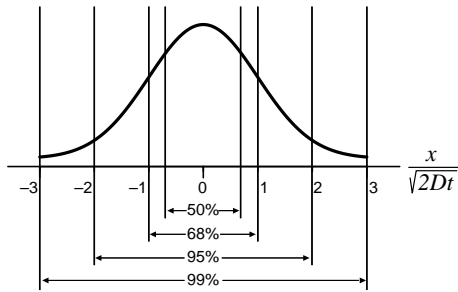


Figure 3.14

How long till half the solute diffuses to  $|x| > x_{1/2}$



$$\frac{x_{1/2}}{\sqrt{2Dt}} \approx \frac{2}{3}$$

$$\frac{2}{3} \sqrt{2Dt} \approx x_{1/2}$$

$$\frac{4}{9} 2Dt \approx x_{1/2}^2$$

$$t \approx \frac{x_{1/2}^2}{D} \equiv t_{1/2}$$

$$c(x,t) = \frac{n_0}{\sqrt{4\pi Dt}} e^{-x^2/(4Dt)}$$

### Importance of Scale

$$t_{1/2} = \frac{x_{1/2}^2}{D} ; D = 10^{-5} \frac{\text{cm}^2}{\text{s}} \text{ for small solutes (e.g., Na}^+) \text{}$$

	$x_{1/2}$	$t_{1/2}$
membrane sized	10 nm	$\frac{1}{10} \mu\text{sec}$
cell sized	10 $\mu\text{m}$	$\frac{1}{10} \text{ sec}$
dime sized	10 mm	$10^5 \text{ sec} \approx 1 \text{ day}$