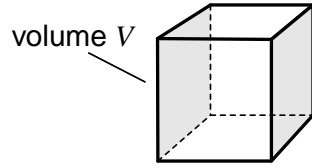


Figure 2.19

Analogous flux laws

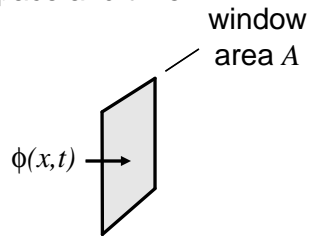
Fourier:	heat flux	α	$\frac{d(\text{temperature})}{dx}$
Ohm:	current	α	$\frac{d(\text{potential})}{dx}$
Fick:	substance flux	α	$\frac{d(\text{concentration})}{dx}$

Concentration at a point
in space and time



$$\text{concentration } c(x,t) = \lim_{V \rightarrow 0} \frac{\text{amount of substance in } V}{V}$$

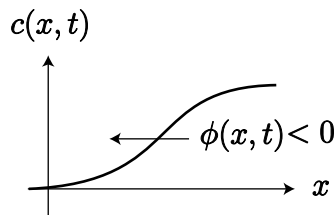
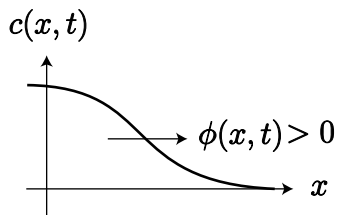
Flux at a point
in space and time



$$\text{flux } \phi(x,t) = \lim_{\substack{A \rightarrow 0 \\ \Delta t \rightarrow 0}} \frac{\text{amount of substance flowing through test window } A \text{ in } \Delta t}{A \Delta t}$$

Fick's First Law

$$\phi(x,t) = -D \frac{\partial c(x,t)}{\partial x}$$



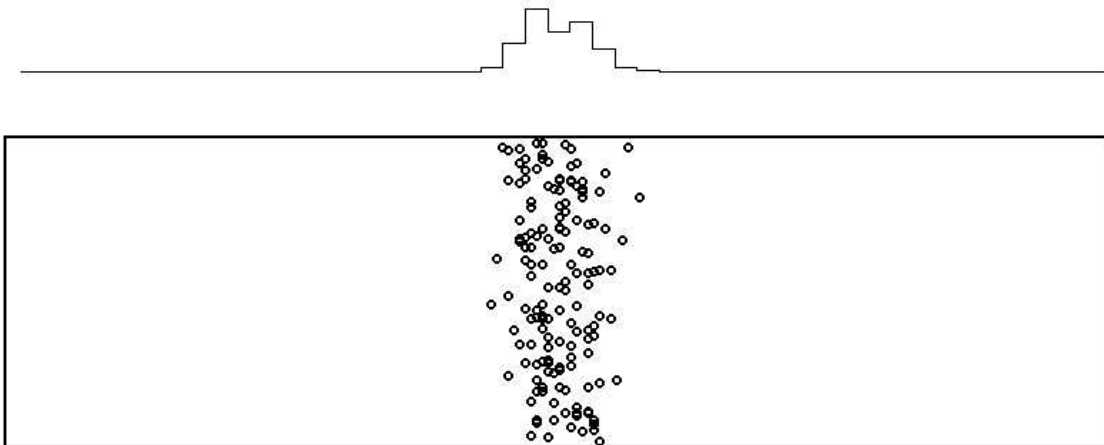
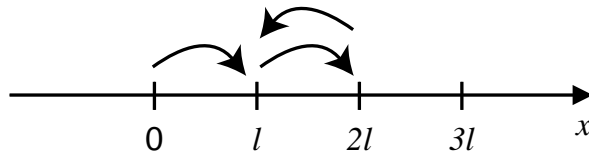
Random Walk Model

- number of solute particles \ll number of solvent particles
- motion of solute determined by collisions with solvent (ignore solute-solute interactions)
- focus on 1 solute particle, assume motions of others are statistically identical

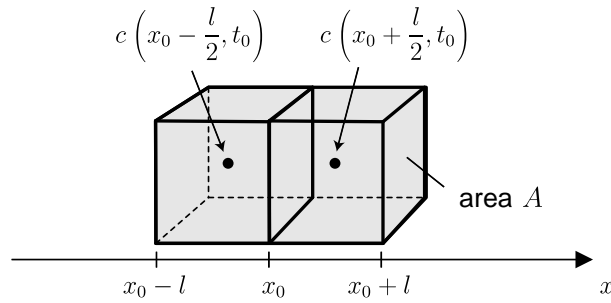
Every τ seconds, solute particle gets hit by solvent particle.

In response, solute particle is equally likely to move $+l$ or $-l$.

τ = mean free time; l = mean free path



Random Walk Model → Fick's First Law



$$\phi^+(x_0, t_0) = \frac{1}{2} \left(\frac{c(x_0 - \frac{l}{2}, t_0) A l}{A \tau} \right) \quad \phi^-(x_0, t_0) = \frac{1}{2} \left(\frac{c(x_0 + \frac{l}{2}, t_0) A l}{A \tau} \right)$$

$$\phi(x_0, t_0) = \frac{l}{2\tau} \left(c(x_0 - \frac{l}{2}, t_0) - c(x_0 + \frac{l}{2}, t_0) \right)$$

$$\phi(x_0, t_0) = -\frac{l^2}{2\tau} \left(\frac{c(x_0 + \frac{l}{2}, t_0) - c(x_0 - \frac{l}{2}, t_0)}{l} \right)$$

$$\phi(x, t) \approx - \left(\frac{l^2}{2\tau} \right) \frac{\partial c(x, t)}{\partial x}$$